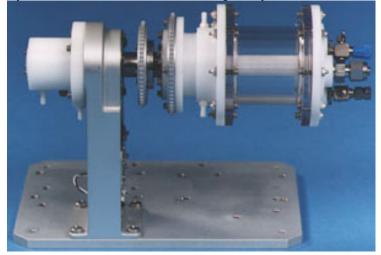
What's Brewing –n- Bioreactors? ChE 1101 – Fall 2005

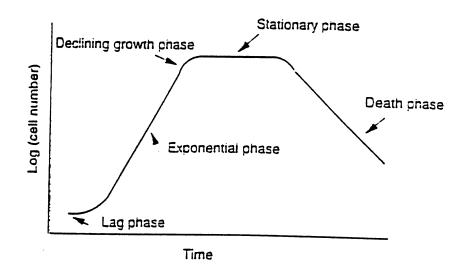
NASA has a number of projects with bioreactors. They are developing them to metabolize human waste (solid, liquid, and gas) on spaceships into breathable air, water, and fertilizer for plants. To read about one of these bioreactors, visit the following website: <u>http://science.msfc.nasa.gov/newhome/br/bioreactor.htm</u>.

A new spin on cell growth

NASA is meeting the challenge with a unique new technology, the rotating wall vessel bioreactor. It spins a fluid medium filled with cells to neutralize most of gravity's effects and encourage cells to grow in a natural manner. The rotating bioreactor was invented by NASA as a model for microgravity effects on cells.



All living populations go through stages of growth, which is depicted in the following graph.



Desktop Experiment Module: Brewing -n- Bioreactors

Think about each stage of growth and match the phase with its definition.

a) Cell number and weight increases rapidly.

- ____1) Lag
 b) The number of forming cells is equal to the number dying cells. Existing cells live off of dwindling food supply and by metabolizing the dying cells. Toxins have built up in the environment and are beginning to interfere with cell function.
 - c) Time it takes for cells to adjust to their environment before beginning to metabolize the food.
 - d) Cells run out of food and the population crashes. Toxins have built up to a lethal level and cells die rapidly.
 - e) Cell growth starts slowing down due to the environment getting crowded and many cells are competing for the food.

We are going to monitor these phases of growth in a bread yeast population in a tiny bioreactor. From the discussion in class, we learned that mass is conserved in every process. In this experiment, we have 2 processes: the cellular process and the bioreactor itself. For the cellular process, it is necessary to understand that cells eat food and excrete waste. Food in our system is going to be sugar, which the cells take in and metabolize to CO₂, ethanol, and hydrocarbons.

3) Declining growth

4) Stationary

5) Death



We are going to monitor CO₂ production by watching how much gas is produced from our reactor.

EXPERIMENT:

Lay out your materials for your bioreactor on your lab mat.

- Step 1. Weigh out 0.2 grams of yeast using weighing paper on the scales and carefully pour this into your bioreactor (erlenmyer flask). Weigh out the amount of sugar your group has been assigned and add that to the bioreactor. **Record these amounts on the Results Page**.
- Step 2. Fill your bioreactor with 50°C (~120°F) water to the 75 mL line (milliliters are the same as cc's or cubic centimeters which are also written cm³). **Record this amount on your Results Page.**
- Step 3. Stir rapidly with your stir bar until all the yeast and sugar is dissolved. Push the stopper with the bubbler in it into the bioreactor flask until it is snug.
- Step 4. Start your timer now.
- Step 5. Remove the cap from the bubbler apparatus on the top of the bioreactor. Using your squirt bottle, fill the bubbler with



water up to the line marked with the arrow in the photograph above.

- Step 6. Place the entire bioreactor (flask and bubbler) inside the temperature bath (beaker). Fill the beaker to the 250 mL mark with warm 50°C (~120°F) water. Place the thermometer in the bath and record the temperature on your Results Page.
- Step 7. Your bioreactor is now together. On your Results Page, record the time of each bubble and the temperature. You may plot the results as you go on the chart provided.

Results Page

Step 1:

Exact weight of yeast, $W_y =$ _____ grams

Weight of sugar, W_s : $W_s =$ _____ grams

Step 2:

Weight of 50°C water added, W_w: Amount of water added = _____ mL Density of water = 1 gram / mL Total weight of water added, **W**_w = _____ **grams**

What is the total mass entering the reactor?

W_T = _____ grams

Step 4:

Bubble #	Timer	Temperature
initial	0.00	
1		
2		
3		

Questions:

Rate of bubble production can be determined from your raw data or from your graphs. This is the number of bubbles produced over a period of time. Locate the 3-minute period of time in which the number of bubbles produced was largest. Use this to calculate the maximum bubble production your biomass showed.

Maximum bubble rate, R_{max} = _____bubbles / sec

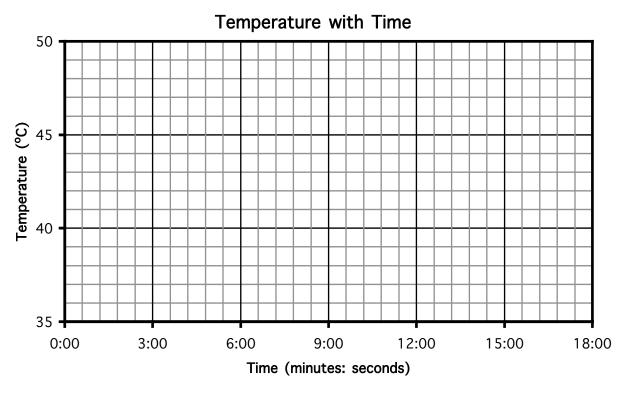
What growth phase was this?_____

Use the same strategy to determine the minimum bubble rate.



What growth phase was this?_____

Graph your temperature results:



What was the rate of cooling (calculate from temperature data)? R_{cool} = $^{\circ}C/sec$

Bioreactor Memo & Results (Due Nov. 3): Write up a 2 page discussion of your results, observations and conclusions. Please include plots of your data. How could a chemical engineer improve the design of these bioreactors to improve results and increase yield?